Molecular Symmetry - Posot groups

Symmetry props. It volcales closely feed to many properties

(D) Electore Structure + MO Structure

(D) Peactivity

(3) Molecular Vi Stratons + EAS

we claracterse symmetry of notcules by placing them outs point La Point Coup - defines a collection of symmetry elevents le print at when at which all symmetry clarent intiged

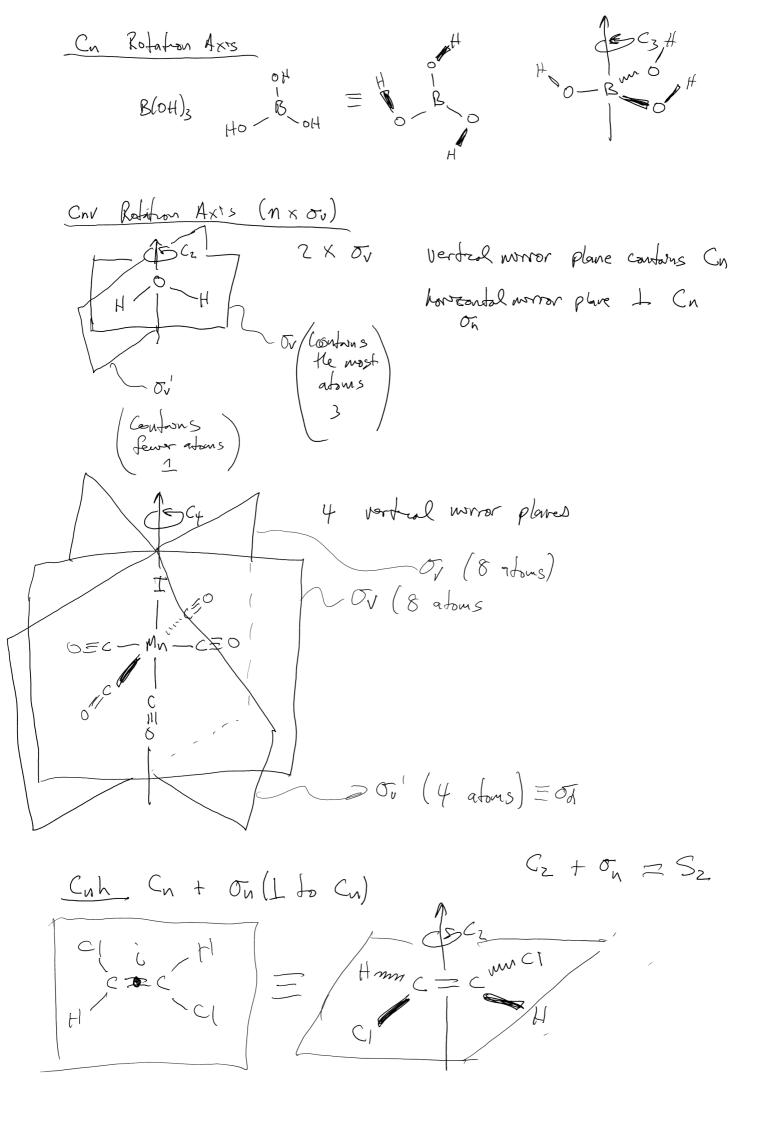
CL (E) No Symmetry

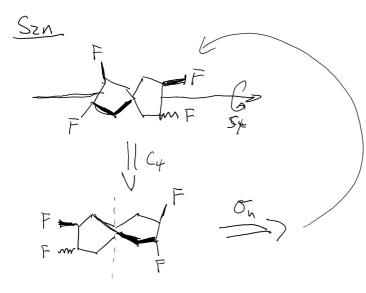
Cs (o) one mirror plane exists

Clam [FPC\Br

Sociz CI

Ci (i) inversion contr de chloro defloro ethane CIMIN S

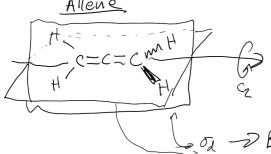




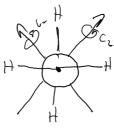
Dn Cn (n I Cz

HIN PHZ = NN (en)

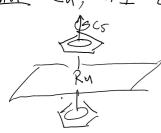
Ond Cn n I Cz nod

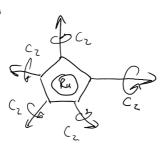


302 - 2 Bisect the I Cz



Only Cn, n L Cz, on



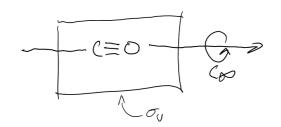


5 = Ge Aromate System



Points of hyler Symmetry

Cov Coo, Ov



Doh Co, co L Cz, on, or or

* Any trear moleule w/ an inversion conser

Td = telahedal

8 C3, 3Cz, 6S4, 602

6 S4, 8S6, 30m, 60d

On = Octobedral C1-W-C1 8C3, GCp, 3C2 C1 C1

Ih = Icosahedol

0, T, I

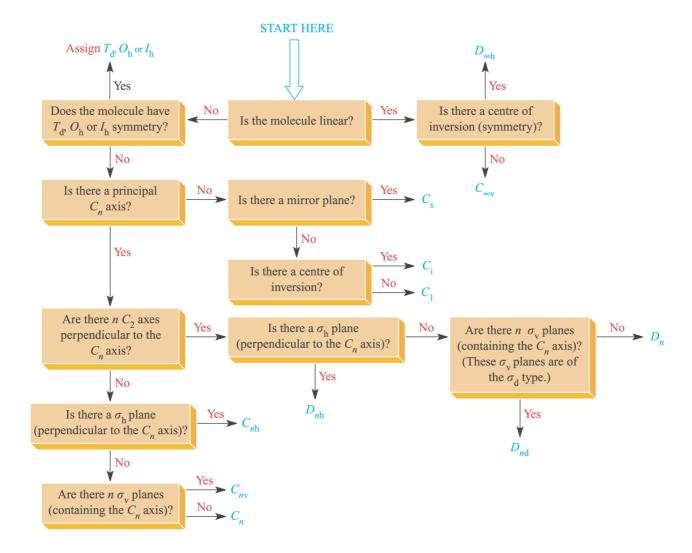
TABLE 4.2 Groups of Low Symmetry

Group	Symmetry	Examples	
C_1	No symmetry other than the identity operation	CHFCIBr	F C Br
C_s	Only one mirror plane	H ₂ C=CClBr	H C=C Cl
C_i	Only an inversion center; few molecular examples	HClBrC—CHClBr (staggered conformation)	Br H C-C Cl Br

TABLE 4.3 Groups of High Symmetry

TABLE 4.3	Groups of High Symmetry		
Group	Description	Examples	
$C_{\infty v}$	These molecules are linear, with an infinite number of rotations and an infinite number of reflection planes containing the rotation axis. They do not have a center of inversion.	C_{∞} H—CI	
$D_{\infty h}$	These molecules are linear, with an infinite number of rotations and an infinite number of reflection planes containing the rotation axis. They also have perpendicular C_2 axes, a perpendicular reflection plane, and an inversion center.	$C_{\infty} O = C_{2}$	
T_d	Most (but not all) molecules in this point group have the familiar tetrahedral geometry. They have four C_3 axes, three C_2 axes, three S_4 axes, and six σ_d planes. They have no C_4 axes.	H C H	
O_{\hbar}	These molecules include those of octahedral structure, although some other geometrical forms, such as the cube, share the same set of symmetry operations. Among their 48 symmetry operations are four C_3 rotations, three C_4 rotations, and an inversion.	F-S-F	
I_h	Icosahedral structures are best recognized by their six C_5 axes, as well as many other symmetry operations—120 in all.	B ₁₂ H ₁₂ ²⁻ with BH at each vertex of an icosahedron	

In addition, there are four other groups, T, T_R, O, and I, which are rarely seen in nature. These groups are discussed at the end of this section.



Point group	Characteristic symmetry elements	Comments
C_{\circ}	E , one σ plane	
C:	E, inversion centre	
C_n	E, one (principal) <i>n</i> -fold axis	
$C_{nv}^{''}$	E, one (principal) <i>n</i> -fold axis, $n \sigma_v$ planes	
$C_{n\mathrm{h}}$	E, one (principal) <i>n</i> -fold axis, one σ_h plane, one S_n -fold axis which is coincident with the C_n axis	The S_n axis necessarily follows from the C_n axis and σ_h plane For $n = 2$, 4 or 6, there is also an inversion centre
$D_{n\mathrm{h}}$	E, one (principal) <i>n</i> -fold axis, n C_2 axes, one σ_h plane, n σ_v planes, one S_n -fold axis	The S_n axis necessarily follows from the C_n axis and σ_h plane For $n = 2$, 4 or 6, there is also an inversion centre
D_{nd}	E, one (principal) <i>n</i> -fold axis, n C_2 axes, n σ_v planes, one S_{2n} -fold axis	For $n = 3$ or 5, there is also an inversion centre
$T_{ m d}$		Tetrahedral
$O_{ m h}$		Octahedral
$I_{ m h}$		Icosahedral

